## KAT 244/3 - Separation Methods

**Course Objective**: To introduce the fundamental principles and theory of separation techniques: sample preparation, gas and liquid chromatography.

Topic	Content	Number of lecture hours	Expected outcome – upon completion of the course, the student should be able to:
Sample     Preparations and     Solvent Extraction	<ul> <li>Distribution coefficient and ratio</li> <li>Justifications for extraction</li> <li>Percent extracted</li> <li>Improving percent extraction</li> <li>Solvent extraction of metals</li> <li>Examples</li> <li>Solid-phase extraction</li> <li>Microwave assisted digestion</li> </ul>	6	<ul> <li>Understand the underlying principles in liquid-liquid extraction.</li> <li>Understand the possible equilibria involved in extraction.</li> <li>Offer ways to increase extraction efficiency.</li> <li>State common applications of extraction.</li> </ul>
2. Chromatography Principles and Theory	<ul> <li>Principles of chromatography</li> <li>Classification of chromatographic techniques</li> <li>Migration rates of solutes</li> <li>Band broadening and column efficiency</li> <li>Variables that affect column efficiency</li> <li>Resolution</li> <li>Quantification</li> </ul>	6	<ul> <li>Explain how solutes are separated on a column.</li> <li>Understand the various types of chromatography.</li> <li>Understand the factors that affect column efficiency.</li> <li>Understand the inter-play of retention time, resolution and capacity on migration.</li> <li>Explain how quantitative and qualitive analysis are performed.</li> </ul>

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Topic	Content	Number of lecture hours	Expected outcome – upon completion of the course, the student should be able to:
3. Gas Chromatography	<ul> <li>Introduction</li> <li>Packed and capillary columns</li> <li>Temperature selection</li> <li>Stationary phases</li> <li>Detectors</li> <li>GC-MS</li> <li>Specialized GC methods</li> </ul>	5	<ul> <li>Distinguish the differences between packed and capillary columns.</li> <li>Understand the properties of stationary phases.</li> <li>Understand the common detectors and their operation.</li> <li>Understand quantitative analysis using internal standards.</li> <li>Understand the benefits to be derived from GC-MS.</li> <li>Discuss the common applications of GC.</li> </ul>
4. Liquid Chromatography	<ul> <li>Principles</li> <li>HPLC components</li> <li>Stationary phases</li> <li>Mobile phase manipulation and optimization strategies</li> <li>Detectors</li> <li>Narrow-bore columns</li> <li>Size exclusion chromatography</li> <li>Ion exchange chromatography</li> <li>Thin layer chromatography</li> </ul>	10	<ul> <li>Understand the components of HPLC.</li> <li>Understand method development – column and solvent selection.</li> <li>Understand the common stationary phases.</li> <li>Understand the advantages of using narrow-bore columns.</li> <li>Discuss the other modes of liquid chromatography.</li> </ul>
5. Electrophoretic Techniques	<ul> <li>Introduction</li> <li>Separation principles</li> <li>Factors affecting solute migration</li> <li>Gel electrophoresis</li> <li>Capillary electrophoresis</li> <li>Electroosmotic mobility</li> <li>Electrophoretic mobility</li> <li>Instrumentation</li> <li>Selected applications</li> </ul>	6	<ul> <li>Understand the principles of electrophoresis.</li> <li>Understand the limitations of gel electrophoresis.</li> <li>Explain the phenomenon of migration.</li> <li>Understand the components of capillary electrophoresis.</li> <li>Discuss the common applications of capillary electrophoresis.</li> </ul>
6. Miscellaneous Techniques	Supercritical fluid chromatography (SFC).	3	Understand the principles of SFC.
TOTAL		36	

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