## KIT 254/2 – Polymer

**Course Objectives** 1) : 2)

To study the polymerization mechanisms and kinetics. To introduce the physical chemical principles and properties of polymers.

Торіс	Content	Number of lecture hours	Expected outcome – upon completion of this course, the student should be able to:
1. Introduction to Polymers	<ul> <li>The importance of polymers to man</li> <li>Basic concepts</li> <li>Classification of polymers</li> <li>The architecture of polymer molecules</li> <li>Polymer morphology</li> <li>Copolymers</li> <li>Plastics</li> <li>Elastomers</li> <li>The effects of temperature on polymers</li> <li>Geometrical arrangement of atoms in polymer molecules</li> <li>Chemical bonds in polymer molecules</li> <li>Crystallinity of polymers</li> </ul>	2	<ul> <li>Differentiate the advantages and disadvantages of polymer materials compared to other substances (for example: metals and wood) on the basis of their properties.</li> <li>Understand the basic terms: polymer, monomer, degree of polymerisation (DP), repeating units, chain molecules.</li> <li>Recognise the basic structures of a polymer: linear, branched, cross-linked and network.</li> <li>Differentiate the molecular arrangement in an amorphous, crystalline and semicrystalline polymers.</li> <li>Understand the difference of morphology on the physical properties of polymers.</li> <li>Understand the difference between homopolymer and copolymer.</li> <li>Recognise the structure of random, alternate, block and graft copolymers.</li> <li>Understand the differences in properties between thermoplastic, thermoset and elastomer.</li> <li>Understand the term glass transition temperature and its relationship with the chemical structure of a polymer.</li> <li>Understand why a polymer behaves as plastic or elastic.</li> <li>Identify several types of intermoleculecular forces and their influence on the physical properties of polymers.</li> </ul>
2. Polymerisation Mechanisms	<ul> <li>Step reaction polymerisation</li> <li>Chain reaction polymerisation</li> <li>Comparison between step and chain reaction polymerisation</li> </ul>	1	<ul> <li>Understand the general characteristics and differences between step reaction and chain reaction polymerisation.</li> </ul>

Торіс	Content	Number of lecture hours	Expected outcome – upon completion of this course, the student should be able to:
3. Step Reaction Polymerisation	<ul> <li>Examples of condensation polymers</li> <li>Kinetics of step polymerisation:         <ul> <li>Self-catalysed</li> <li>External-catalysed</li> </ul> </li> <li>Quantitative aspects of step polymerisation</li> <li>Molecular weight control</li> </ul>	3	<ul> <li>Recognise the structure of condensation polymers and their monomers.</li> <li>Derive the kinetic equations for self-catalysed and external-catalysed step polymerisation.</li> <li>Understand the effect of both types of catalysis to DP and molecular weight.</li> <li>Understand the use of the Carothers equation.</li> <li>Understand the methods to control molecular weight and use the appropriate equations in calculations.</li> </ul>
4. Radical Chain Polymerisation	<ul> <li>Examples of chain polymers</li> <li>Main characteristics</li> <li>Free radical initiators</li> <li>Mechanism of polymerisation</li> <li>Kinetics of chain polymerisation:         <ul> <li>system with initiator</li> <li>system without initiator</li> </ul> </li> <li>Kinetic chain length</li> </ul>	3	<ul> <li>Write the three-step mechanism for a vinyl monomer.</li> <li>Understand the main characteristics of a radical chain polymerisation.</li> <li>Recognise several examples of free radical initiators and know their functions.</li> <li>Understand the schematic mechanism of radical chain polymerisation.</li> <li>Derive the kinetic equations for each step of the mechanism.</li> <li>Understand the steady state concept.</li> <li>Derive equations of the rate of polymerisation for a system with initiator and without initiators.</li> <li>Understand the term 'kinetic chain length'.</li> <li>Derive the equation for kinetic chain length.</li> </ul>
5. Chain Transfer Reaction	<ul> <li>Types of chain transfer reactions</li> <li>The effects of chain transfer</li> <li>Kinetics of a chain transfer reaction</li> <li>Control of molecular weight by chain transfer</li> </ul>	3	<ul> <li>Understand the concept of a chain transfer and the effects on the molecular weight of polymers.</li> <li>Derive kinetic equations that relate DP with the rate of polymerisation and the rate of a polymer formation.</li> <li>Understand a method of molecular weight control by a kinetic chain transfer.</li> </ul>

Торіс	Content	Number of lecture hours	Expected outcome – upon completion of this course, the student should be able to:
6. Ionic Polymerisation	<ul> <li>Reaction mechanisms in ionic (cationic and anionic) polymerisation of vinyl monomer</li> <li>Kinetics – polymerisation kinetics of anion and cation processes, influence of solvent, counter-ion and temperature</li> <li>Crosslinking formation in polymers and their applications in industries</li> </ul>	3	<ul> <li>Choose monomer that can be polymerised through anionic and/or cationic polymerisation techniques.</li> <li>Write polymerisation reaction mechanisms involved in anionic and cationic systems.</li> <li>Illustrate rate and degree of polymerisation from kinetic schemes.</li> <li>Relate factors that influence polymerisation with kinetic equations.</li> <li>Understand and write crosslinking reaction mechanisms of a given polymer and can select special routes to crosslink different polymers.</li> <li>Relate crosslinking behaviour with polymer based products.</li> </ul>
7. Polymers Characterisation	<ul> <li>Polymer molar mass – molar mass distributions; average molar masses</li> <li>Determination techniques on molar mass: gel permeation (size exclusion) chromatography and membrane and vapour pressure osmometry</li> <li>Determination of polymer viscosity: dilute solution viscometry ; intrinsic viscosity and the Mark-Houwink relationship for M<sub>v</sub></li> </ul>	4	<ul> <li>Calculate molar mass using an equation and identify their distribution in the polymer matrix.</li> <li>Use the instruments to measure the molar mass of polymers.</li> <li>Indicate some equipments to characterise polymers.</li> </ul>

Торіс	Content	Number of lecture hours	Expected outcome – upon completion of this course, the student should be able to:
8. Physical and Mechanical Behaviour of Polymers	<ul> <li>Physical : basic concepts of glass transition temperature, melting behaviour in polymers and factors effecting melting and glass transition temperatures</li> <li>Mechanical : basic concepts of mechanical behaviour in polymers and composites. Experimental studies on polymer films and fibres. Yields criteria e.g. determination of stress, strain, modulus,UTS and elongation at break</li> <li>Equipments and techniques to determine glass transition temperature, melting temperature, mechanical properties.</li> </ul>	5	<ul> <li>Understand a basic knowledge of the physical properties of solid polymeric materials and how properties and chemical and materials structures are related.</li> <li>Calculate modulus, UTS, elongation at break from tensile tests and 3 point bending data.</li> <li>Choose equipments for the estimation of physical and mechanical properties of polymers.</li> </ul>
TOTAL		24	

UUTTHH