

## KIT 257/3 – Material Chemistry

**Course Objective** : To introduce the chemical aspects and properties of materials and their applications – metals and alloys, polymers, ceramics and composites.

Topic	Content	Number of lecture hours	Expected outcome - upon completion of this course, the student should be able to:
1. Introduction	<ul style="list-style-type: none"> <li>Materials science and engineering</li> <li>The importance of materials in life and industry</li> <li>Classification of materials</li> <li>Different levels of structure in materials</li> <li>Relation among structure, processing and properties</li> </ul>	1	<ul style="list-style-type: none"> <li>Understand the importance of materials in life and the role of materials scientists in the future development of industry.</li> <li>Classify the solid materials and cite the distinctive chemical features of each.</li> <li>Understand the relation among structure, processing and properties of materials.</li> </ul>
2. Atomic Structure and Chemical Bonding	<ul style="list-style-type: none"> <li>Atomic structure</li> <li>Chemical bonding and binding energy</li> <li>Properties from bonding</li> <li>Unit cells</li> <li>Crystal structure of solids</li> <li>Crystallographic directions and planes</li> <li>Determination of crystal structures</li> </ul>	3	<ul style="list-style-type: none"> <li>Understand the fundamental concepts of chemical bonding.</li> <li>Describe the type of bonds and note which materials exhibit those bonding types.</li> <li>Understand the assembly of atoms in solid structures and the classification of crystals (crystal systems).</li> <li>Specify and sketch the crystallographic directions and planes within a unit cell.</li> <li>Distinguish between single crystals and polycrystalline materials.</li> </ul>
3. Imperfection in Solid	<ul style="list-style-type: none"> <li>Types of imperfections/defects</li> <li>Defects in ceramic structures</li> <li>Defects in alloys</li> <li>Characterization of defects</li> </ul>	2	<ul style="list-style-type: none"> <li>Describe the types of defects in solids.</li> <li>Understand the criteria of equilibrium in solid structures.</li> <li>Calculate the equilibrium number of vacancies in a material.</li> <li>Understand the techniques for characterizing defects.</li> </ul>
4. Diffusion in Solids	<ul style="list-style-type: none"> <li>Types of diffusion</li> <li>Diffusion mechanisms</li> <li>Factors that influence diffusion</li> <li>Effects of diffusion on the structure and properties of materials</li> </ul>	2	<ul style="list-style-type: none"> <li>Understand and describe the mechanisms of diffusion within solid materials.</li> <li>Distinguish between steady-state and non steady –state diffusions.</li> <li>Calculate the diffusion coefficient and activation energy for some materials under a given temperature.</li> <li>Understand the factors that influence diffusion.</li> </ul>

Topic	Content	Number of lecture hours	Expected outcome - upon completion of this course, the student should be able to:
5. Ceramics	<ul style="list-style-type: none"> <li>• Basic categories of ceramics</li> <li>• General properties of ceramic materials</li> <li>• Structure of ceramics</li> <li>• Silicates and glasses</li> <li>• New and modern ceramics</li> <li>• Biodegradable and bioactive ceramics</li> <li>• Applications of ceramic materials</li> </ul>	3	<ul style="list-style-type: none"> <li>• Understand the basic categories of ceramics.</li> <li>• Understand the types of bonding, structure and defects in ceramic structures.</li> <li>• Understand the importance of silicates in ceramic materials.</li> <li>• Relate ceramic structures with its properties and applications.</li> <li>• Understand the features of modern ceramic materials.</li> </ul>
6. Polymers	<ul style="list-style-type: none"> <li>• Structures and types of polymers</li> <li>• Molecular weight, degree of polymerisation</li> <li>• Amorphous and crystalline polymers</li> <li>• Synthesis of polymers</li> <li>• Phase transition of polymers</li> </ul>	2	<ul style="list-style-type: none"> <li>• Understand the basic microstructural features of polymers.</li> <li>• Understand the types of polymers, properties and applications.</li> <li>• Calculate the number-average and weight-average molecular weights and degree of polymerisation for specified polymers.</li> <li>• Describe the crystalline state of polymers.</li> <li>• Understand the basic route for polymer synthesis.</li> <li>• Understand the phase transition in polymers when heated.</li> </ul>
7. Metals and Alloys	<ul style="list-style-type: none"> <li>• Classification of metals and alloys</li> <li>• Bonding in metals</li> <li>• Metallic structure</li> <li>• Phase diagram of metals (iron)</li> </ul>	1	<ul style="list-style-type: none"> <li>• Understand the bonding in metals and alloys.</li> <li>• Understand the conduction and insulation in materials.</li> <li>• Understand the structures of a metal and its phase diagram.</li> </ul>
8. Composites	<ul style="list-style-type: none"> <li>• General requirements for composites</li> <li>• Types of composites</li> <li>• Forms of matrices and reinforcement phases</li> <li>• Concrete and hybrid composites</li> <li>• Benefits and applications of composite materials</li> </ul>	3	<ul style="list-style-type: none"> <li>• Understand the requirements of composite materials.</li> <li>• Describe the types of composites according to the matrix and reinforcement phases.</li> <li>• Understand the types of interfacial bonding in composites.</li> <li>• Understand the factors that influence the performance and properties of composites.</li> <li>• Understand the features of modern composite materials and its applications.</li> </ul>

Topic	Content	Number of lecture hours	Expected outcome - upon completion of this course, the student should be able to:
9. Phase Diagrams	<ul style="list-style-type: none"> <li>• Basic concepts of a phase equilibrium</li> <li>• Classification of phase diagrams</li> <li>• Interpretation of phase diagrams</li> <li>• The Lever Rule</li> <li>• Development of microstructures</li> <li>• Phase transformations</li> <li>• Factors that influence the phase transformation</li> </ul>	6	<ul style="list-style-type: none"> <li>• Understand the parameters of a phase diagram.</li> <li>• Determine the type of phases, the compositions and the mass fractions of phases for a given phase diagram.</li> <li>• Understand and use the Lever Rule for characterising a given phase diagram.</li> <li>• Understand the development of microstructure in a given phase diagram when heated or cooled.</li> <li>• Describe the phase transformation and relate the process with the microstructure of materials.</li> </ul>
10. Properties of Materials	<ul style="list-style-type: none"> <li>• Mechanical properties: Stress, strain, elastic and plastic behaviour, strength, hardness, ductility and toughness</li> <li>• Electrical properties: conductivity, electron energy bands, electron mobility, semiconductors and dielectric materials</li> <li>• Magnetic properties: magnetic force, magnetic field, classification of magnetic materials and its magnetic properties</li> <li>• Thermal properties: heat capacity, thermal conductivity, thermal expansion and thermal stress/shock</li> <li>• Optical properties: reflection, refraction, absorption and transmission, colour and fiber optic</li> </ul>	8	<ul style="list-style-type: none"> <li>• Understand the relationship among material properties, structure and performance for the classes of engineering solids (metals, polymers, ceramics, semiconductors and composites).</li> <li>• Understand and describe mechanical properties such as stress, strain, strength, hardness, ductility and toughness.</li> <li>• Understand and describe electrical properties such as conductivity, resistivity, electron mobility, conduction in semiconductors and dielectric materials.</li> <li>• Understand and describe magnetic properties such as magnetic force, magnetic field and magnetic susceptibility. Differentiate the types of magnetic materials.</li> <li>• Understand and describe thermal properties such as heat capacity, thermal conductivity, thermal expansion and thermal stress/shock.</li> <li>• Understand and describe optical properties such reflection, refraction, absorption and transmission, colour and the fiber optic.</li> </ul>
11. Corrosion and Degradation of Materials	<ul style="list-style-type: none"> <li>• Corrosion of metals: corrosion reaction and corrosion rate, factors that influence the corrosion, forms of corrosion, corrosion protections</li> <li>• Degradation of polymer: swelling, dissolution, bond rupture and weathering.</li> </ul>	5	<ul style="list-style-type: none"> <li>• Describe the corrosion reaction and the influence of environment conditions on the corrosion rate.</li> <li>• Describe the nature of corrosion process, forms of corrosion and methods of corrosion prevention.</li> <li>• Explain why ceramic materials are very resistant to corrosion.</li> <li>• Describe the common process of degradation in polymers.</li> </ul>
<b>TOTAL</b>		<b>36</b>	