

KOE 322/3 – Natural Products Chemistry

- Course Objectives :
- 1) To introduce the synthetic pathway of natural products such as acetate, shikimate, mevalonate and alkaloid.
 - 2) To learn the synthetic technique reported by researcher in the field of natural products chemistry.

Topic	Content	Number of lecture hours	Expected outcome – upon completion of this course, the student should be able to:
1. Introduction about natural products	<ul style="list-style-type: none">• Types of natural products and organisms that produce them• Comparison with the synthetic products• Plants natural product.• Organic catalyst (enzymes)	2	<ul style="list-style-type: none">• Discuss what we mean by the term of natural products.• Familiar with the kinds of organisms that make them.• Understand the uses of enzymes by the microbe to make a chemical.
	<ul style="list-style-type: none">• Extraction of volatile components• Extraction of non volatile components• Methods of isolation (TLC, Column chromatography, GC, GC-MS and HPLC• Structure elucidation (eg: flavonoid)	3	<ul style="list-style-type: none">• Understand the uses of solvents in the extraction• Familiar with the terms “ydrodistillation and head space”• Suggest the suitable methods for the identification and isolation.• Characterize the structure.

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2. Outline of the biosynthesis of natural secondary metabolites	<ul style="list-style-type: none"> • The main building blocks and the basic construction mechanisms employed in the biosynthesis of natural products. • Biological alkylation reactions, Wagner-Meerwein Rearrangements, Aldol addition and Claisen condensation, Diels-Alder reactions, Schiff base formation, Mannich reaction, transamination, decarboxylation reactions, phenolic oxidative coupling, redox reactions, oxidative cleavage of aromatic rings and glycosylation reactions • Making new carbon-carbon bonds, malonic ester synthesis, the acetoacetic ester synthesis, acylation, Michael reaction • A biological aldol condensation, Robinson annulation • Designing a synthesis, controlling stereochemistry, hydroxylation of alkene, oxidative cleavage of alkenes, functional group interconversion • Biological oxidation-reductions, oxidation of hydroquinone and reduction of quinines 	7	<ul style="list-style-type: none"> • Relate many of these reactions with the many fundamental principles of organic chemistry.

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3. The acetate pathway	<ul style="list-style-type: none"> • The biosynthesis of fatty acids, prostaglandins, and polyketides. Cyclization of polyketides. 	3	<ul style="list-style-type: none"> • Discuss the biosynthesis of saturated and unsaturated fatty acids, and prostaglandins. • Rationalize the cyclization of polyketides in terms of Aldol and Claisen condensations. • Suggest biosynthetic route to simple polyketides.
4. The shikimate pathway	<ul style="list-style-type: none"> • The role of shikimic acid in the formation of aromatic amino acids, benzoic acids and cinnamic acids, along with further modifications leading to lignans and lignin, polypropenes and coumarins. • Combinations of shikimate and acetate pathways in the biosynthesis of styrylpyrones, flavonols, isoflavonols and stibenes. 	3	<ul style="list-style-type: none"> • Discuss the biosynthesis of certain aromatic amino acids and phenylpropanoids. • Suggest biosynthetic route to phenylpropanoids.
5. The mevalonate pathway	<ul style="list-style-type: none"> • Classification of terpenes according to the number of isoprenoid units incorporated. • The mevalonate pathway leading to terpenes. Steroids as examples of modified triterpenes. • The stereochemistry of steroids. 	4	<ul style="list-style-type: none"> • Discuss the biosynthesis of terpenes via the mevalonate pathway. • Rationalize terpenoid structures through extensive use of carbocation mechanisms and Wagner-Meerwein rearrangements. • Recognize terpenoids among natural products. • Suggest biosynthetic route to simple terpenoids.

Topic	Content	Number of lecture hours	Expected outcome – upon completion of this course, the student should be able to:
6. Biosynthesis of alkaloids	<ul style="list-style-type: none"> • Types of alkaloids • Some examples of alkaloid biosynthesis, such as alkaloids derived from ornithine, lysine, nicotinic acid, tyrosine, tryptophan, phenylalanine and histidine 	3	<ul style="list-style-type: none"> • Recognize the fact that the major source for the biosynthesis of alkaloids is just a few amino acids, and a major reaction (Mannich reaction).
7. Synthetic work	<ul style="list-style-type: none"> • Selected total synthesis of natural products • Total synthesis of reserpine • Total synthesis of monensin • The synthesis of marine natural product, neumarinone • The synthesis of warburganal from natural product, confertifolin • Synthesis of natural products analogue - The synthesis of carnosol derivatives • The synthesis of difluorinated analogue of shikimic acid - The synthesis of an analogue of strongylodiols 	7	<ul style="list-style-type: none"> • Understand the important of total synthesis. • Understand the important of the synthesis of natural products analogue.
8. Current developments and future directions	<ul style="list-style-type: none"> • The current areas in the field of natural products chemistry by discussing recent publications • Drugs discovery and applications • The current topic in the synthesis of natural products such as solid phase synthesis, combinatorial synthesis, catalysis, the uses of enzyme in synthesis, chiral and stereoselective synthesis) 	4	<ul style="list-style-type: none"> • Create student awareness on current ongoing research.
TOTAL		36	