

## KUT 307/2 - Chemistry Practical VII (Inorganic and Analytical)

- Course Objectives:**
- 1) To acquire skills in synthesis, isolation and characterization of inorganic compounds.
  - 2) To acquire skills in various analytical techniques including spectrophotometric titration, flame emission and atomic absorption spectrometry and gas chromatography.

Experiment title	Content	Number of laboratory hours	Expected outcome – upon completion of these experiments, the student should be able to:
<b>Inorganic</b>			
1. Organotin compounds: Formation of organotin adduct and outer sphere complexes by the reaction of $\text{Ph}_2\text{SnCl}_2$ with Schiff Bases.		6	<ul style="list-style-type: none"> <li>• Synthesize organotin adduct and outer-sphere complexes.</li> <li>• Purify the products (adduct and complexes) by recrystallization.</li> <li>• Characterize the products (adduct and complexes) based on melting point and spectroscopic data (FT-IR and <math>^1\text{H}</math> FT-NMR).</li> </ul>
2. Synthesis of $[\text{Co}(\text{CH}_3)(\text{DMG})_2\text{Py}]$ .		6	<ul style="list-style-type: none"> <li>• Synthesize organocobalt complex <math>[\text{Co}(\text{CH}_3)(\text{DMG})_2\text{Py}]</math> under inert atmosphere (<math>\text{N}_2</math>).</li> <li>• Isolate and dry the product (complex) under reduced pressure.</li> <li>• Characterize the complex based on melting point, elemental analysis (C &amp; H) and spectroscopic data (FT-IR and <math>^1\text{H}</math> FT-NMR).</li> </ul>

Experiment title	Content	Number of laboratory hours	Expected outcome – upon completion of these experiments, the student should be able to:
3. Preparation and characterization of Ferrocene [bis(cyclopentadienyl)ferrocene] [bis(cyclopentadienyl)ferrocene(II)]		6	<ul style="list-style-type: none"> <li>• Distill (purify) starting material such as cyclopentadiene under inert atmosphere (N<sub>2</sub>).</li> <li>• Synthesize organoiron complex, ferrocene.</li> <li>• Purify ferrocene via sublimation.</li> <li>• Characterize ferrocene based on elemental analysis (C &amp; H) and spectroscopic data (UV, FT-IR and <sup>1</sup>H FT-NMR).</li> </ul>

Experiment title	Content	Number of laboratory hours	Expected outcome – upon completion of these experiments, the student should be able to:
<b>Analytical</b>			
1. Visible spectrophotometry	<ul style="list-style-type: none"> <li>• Spectrophotometric Titrations</li> </ul>	3	<ul style="list-style-type: none"> <li>• Perform various photometric titrations.</li> </ul>
2. Spectrofluorimetry:	<ul style="list-style-type: none"> <li>• Determination of Pharmaceuticals (Acetylsalicylic (ASA) and Salicylic (SA) Acids) by Fluorometric Method</li> </ul>	3	<ul style="list-style-type: none"> <li>• Perform quantitative analysis using spectrofluorimetry.</li> </ul>
3. Flame Emission Spectrometry:	<ul style="list-style-type: none"> <li>• Flame Emission Spectrometric Determination of Sodium.</li> </ul>	3	<ul style="list-style-type: none"> <li>• Determine sodium using an internal standard method.</li> </ul>
4. Atomic Absorption Spectrometry:	<ul style="list-style-type: none"> <li>• Atomic absorption spectrometry</li> </ul>	3	<ul style="list-style-type: none"> <li>• Understand the effect of organic solvents and strong ligands in metal determination using atomic absorption spectrometry.</li> </ul>
5. Gas chromatography	<ul style="list-style-type: none"> <li>• Quantitative Gas Chromatographic Analysis of a Multi Component Mixture.</li> </ul>	3	<ul style="list-style-type: none"> <li>• Separate and determine the amount of volatile components (cyclohexane and ethylbenzene) using the internal standard gas chromatographic technique.</li> </ul>
	<b>TOTAL</b>	<b>33</b>	